

**PENGARUH BESAR *VOLTAGE* PADA PROSES *ANODIZING* TERHADAP
LAJU DEGRADASI KOMPOSIT *BIODEGRADABLE* Mg-xC_{Ap} R4**

SKRIPSI

Diajukan untuk memenuhi salah satu dari syarat untuk memperoleh Gelar Sarjana
Sains Program Studi Fisika Kelompok Bidang Kajian Fisika Material



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PENGARUH BESAR *VOLTAGE* PADA PROSES *ANODIZING* TERHADAP
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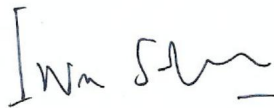
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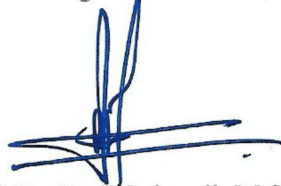


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Dengan ini saya menyatakan bahwa skripsi dengan judul “**PENGARUH BESAR VOLTAGE PADA PROSES ANODIZING TERHADAP LAJU DEGRADASI KOMPOSIT *BIODEGRADABLE* Mg-xCAp R4**” ini beserta seluruh isinya adalah benar-benar karya saya sendiri. Saya tidak melakukan penjiplakan atau pengutipan dengan cara yang tidak sesuai dengan etika ilmu yang berlaku dalam masyarakat keilmuan. Atas pernyataan ini, saya siap menanggung risiko/sanksi apabila di kemudian hari ditemukan adanya pelanggaran etika keilmuan atau ada klaim dari pihak lain terhadap keaslian karya saya ini.

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ABSTRAK

Implan *biodegradable* merupakan salah satu inovasi yang dapat menggantikan peran implan bio-inert dalam aplikasi implant non-permanen untuk luka tulang traumatik karena penggunaannya dianggap lebih efisien dari segi proses penyembuhan dan biayanya. Salah satu material *biodegradable* yang berpotensi untuk dimanfaatkan sebagai bahan dasar implant non-permanen adalah magnesium karena merupakan jenis biometal yang karakteristik mekaniknya hampir sebanding dengan tulang. Namun, laju degradasi magnesium yang tinggi menjadi salah satu halangan dalam proses realisasi pengaplikasian magnesium sebagai bahan dasar implant tulang. Berbagai cara, seperti pembentukan magnesium alloy, struktur komposit berbasis magnesium (salah satunya adalah Mg-CAp), dan *surface treatment* telah dilakukan untuk mengatasi permasalahan tersebut. Metode *anodizing* adalah salah satu metode *surface treatment* yang juga dapat dipilih sebagai langkah lebih lanjut untuk meningkatkan kualitas laju degradasi implan berbasis magnesium. Penelitian ini berfokus kepada pengaruh pemberian tegangan terhadap laju degradasi serta ketebalan dan unsur lapisan anodik yang terbentuk pada permukaan komposit Mg-xCAp ($x = 0, 5, 10, \text{ dan } 15 \text{ wt\%}$). Proses *anodizing* dilakukan selama 15 menit dengan tiga variasi tegangan berbeda (10, 15, dan 20 V). Laju degradasi sampel diperoleh melalui proses pengujian korosi elektrokimia (OCP, EIS, dan Tafel), sedangkan ketebalan dan unsur lapisan anodik diamati melalui uji SEM-EDX-Mapping. Hasil penelitian menunjukkan bahwa peningkatan tegangan *anodizing* dapat meningkatkan ketahanan korosi serta ketebalan dan konsentrasi unsur oksigen yang terkandung pada lapisan anodik. Akan tetapi, peningkatan besar tegangan ternyata juga dapat menimbulkan retakan pada lapisan anodik.

Kata Kunci: Implan Tulang; Komposit Mg-CAp; Korosi; Anodisasi; Lapisan Film Anodik Oksida.

ABSTRACT

Biodegradable implants are one of many innovations that can be utilized as a promising alternative to bio-inert implants in developing temporary implants for bone injury treatment because their application is considered more effective in terms of the healing process and cost. Magnesium, as a biodegradable metal, holds significant potential as a base material for temporary implants due to its mechanical properties, which closely resemble those of bone. However, the high degradation rate of magnesium has become one drawback to its application. Several approaches, including the development of magnesium alloys, magnesium-based composites (such as Mg-CAp), and various surface treatments (such as anodizing), can be employed to mitigate its drawbacks. This research focuses on observing the influence of voltage used in anodizing treatment on the degradation rate of anodized Mg-CAp, thickness, and elements of the anodic oxide film formed on the surface of Mg-xCAp (x = 0, 5, 10, and 15 wt%). The anodizing process was carried out for 15 minutes with three different voltage variations (10, 15, and 20 V). The sample's degradation rate was determined using electrochemical corrosion tests (OCP, EIS, and Tafel), while the thickness and elemental composition of the anodic film were analyzed using SEM-EDX-mapping. The results show that increasing the anodizing voltage can improve corrosion resistance and the thickness and concentration of oxygen elements in the anodic film. However, it has also been found that increasing the voltage can give rise to the formation of cracks in the anodic film.

Keyword: Bone Implants; Mg-xCAp composite; Corrosion; Anodization; Anodic Oxide Film.

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DAFTAR PUSTAKA

- Abdo, H. S., Seikh, A. H., Mohammed, J. A., Luqman, M., Ragab, S. A., & Almotairy, S. M. (2020). Influence of Chloride Ions on Electrochemical Corrosion Behavior of Dual-Phase Steel over Conventional Rebar in Pore Solution. *Applied Sciences*, *10*(13), 4568. <https://doi.org/10.3390/app10134568>
- Abdurahman, R., Eliza, R., Manggala, A., Suci Ningsih, A., & Effendy A, S. (2021). PRODUKSI GAS HIDROGEN BERDASARKAN PENGARUH LUAS PENAMPANG TERHADAP KONSENTRASI LARUTAN ELEKTROLIT DAN SUPPLAI ARUS DENGAN METODE ELEKTROLISIS. *Jurnal Pendidikan dan Teknologi Indonesia*, *1*(11), 447–451. <https://doi.org/10.52436/1.jpti.103>
- Agarwal, S., Curtin, J., Duffy, B., & Jaiswal, S. (2016). Biodegradable magnesium alloys for orthopaedic applications: A review on corrosion, biocompatibility and surface modifications. *Materials Science and Engineering: C*, *68*, 948–963. <https://doi.org/10.1016/j.msec.2016.06.020>
- Alabbasi, A., Mehjabeen, A., Kannan, M. B., Ye, Q., & Blawert, C. (2014). Biodegradable polymer for sealing porous PEO layer on pure magnesium: An in vitro degradation study. *Applied Surface Science*, *301*, 463–467. <https://doi.org/10.1016/j.apsusc.2014.02.100>
- Alamri, A. H. (2020). Localized corrosion and mitigation approach of steel materials used in oil and gas pipelines – An overview. Dalam *Engineering Failure Analysis* (Vol. 116). Elsevier Ltd. <https://doi.org/10.1016/j.engfailanal.2020.104735>
- Amukarimi, S., & Mozafari, M. (2021). Biodegradable magnesium-based biomaterials: An overview of challenges and opportunities. Dalam *MedComm* (Vol. 2, Nomor 2, hlm. 123–144). John Wiley and Sons Inc. <https://doi.org/10.1002/mco2.59>
- Archunan, M. W., & Petronis, S. (2021). Bone Grafts in Trauma and Orthopaedics. *Cureus*. <https://doi.org/10.7759/cureus.17705>
- Arcos, D., & Vallet-Regí, M. (2020). Substituted hydroxyapatite coatings of bone implants. Dalam *Journal of Materials Chemistry B* (Vol. 8, Nomor 9, hlm. 1781–1800). Royal Society of Chemistry. <https://doi.org/10.1039/c9tb02710f>
- Arunnelliappan, T., Baskaran, S., Arun, S., & Prithvirajan, R. (2021). Corrosion behaviour of detonation gun sprayed cermet coatings on AA5083. *Surface Engineering*, *37*(2), 263–270. <https://doi.org/10.1080/02670844.2020.1807096>

- Atrens, A., Liu, M., & Zainal Abidin, N. I. (2011). Corrosion mechanism applicable to biodegradable magnesium implants. *Materials Science and Engineering: B*, *176*(20), 1609–1636. <https://doi.org/10.1016/j.mseb.2010.12.017>
- Buchanan, R. A., & Stansbury, E. E. (t.t.). *ELECTROCHEMICAL CORROSION*.
- Callister, W. D. (2007). *Materials science and engineering : an introduction*. John Wiley & Sons.
- Campo, R. del, Savoini, B., Muñoz, A., Monge, M. A., & Garcés, G. (2014). Mechanical properties and corrosion behavior of Mg-HAP composites. *Journal of the Mechanical Behavior of Biomedical Materials*, *39*, 238–246. <https://doi.org/10.1016/j.jmbbm.2014.07.014>
- Chai, L., Yu, X., Yang, Z., Wang, Y., & Okido, M. (2008a). Anodizing of magnesium alloy AZ31 in alkaline solutions with silicate under continuous sparking. *Corrosion Science*, *50*(12), 3274–3279. <https://doi.org/10.1016/j.corsci.2008.08.038>
- Chai, L., Yu, X., Yang, Z., Wang, Y., & Okido, M. (2008b). Anodizing of magnesium alloy AZ31 in alkaline solutions with silicate under continuous sparking. *Corrosion Science*, *50*(12), 3274–3279. <https://doi.org/10.1016/j.corsci.2008.08.038>
- Chakik, F. ezzahra, Kaddami, M., & Mikou, M. (2017). Effect of operating parameters on hydrogen production by electrolysis of water. *International Journal of Hydrogen Energy*, *42*(40), 25550–25557. <https://doi.org/10.1016/j.ijhydene.2017.07.015>
- Cipriano, A. F., Lin, J., Miller, C., Lin, A., Cortez Alcaraz, M. C., Soria, P., & Liu, H. (2017). Anodization of magnesium for biomedical applications – Processing, characterization, degradation and cytocompatibility. *Acta Biomaterialia*, *62*, 397–417. <https://doi.org/10.1016/j.actbio.2017.08.017>
- Dagdag, O., Safi, Z., Wazzan, N., Erramli, H., Guo, L., Mkadmh, A. M., Verma, C., Ebenso, E. E., El Gana, L., & El Harfi, A. (2020). Highly functionalized epoxy macromolecule as an anti-corrosive material for carbon steel: Computational (DFT, MDS), surface (SEM-EDS) and electrochemical (OCP, PDP, EIS) studies. *Journal of Molecular Liquids*, *302*, 112535. <https://doi.org/10.1016/j.molliq.2020.112535>
- Daniyal, M., & Akhtar, S. (2020). Corrosion assessment and control techniques for reinforced concrete structures: a review. *Journal of Building Pathology and Rehabilitation*, *5*(1), 1. <https://doi.org/10.1007/s41024-019-0067-3>
- de Oliveira, M. C. L., Pereira, V. S. M., Correa, O. V., & Antunes, R. A. (2014). Corrosion Performance of Anodized AZ91D Magnesium Alloy: Effect of the

- Anodizing Potential on the Film Structure and Corrosion Behavior. *Journal of Materials Engineering and Performance*, 23(2), 593–603. <https://doi.org/10.1007/s11665-013-0755-0>
- Ding, W. (2016). Opportunities and challenges for the biodegradable magnesium alloys as next-generation biomaterials. *Regenerative Biomaterials*, 3(2), 79–86. <https://doi.org/10.1093/rb/rbw003>
- Eliaz, N. (2019). Corrosion of Metallic Biomaterials: A Review. *Materials*, 12(3), 407. <https://doi.org/10.3390/ma12030407>
- Friesenbichler, J., Maurer-Ertl, W., Sadoghi, P., Pirker-Fruehauf, U., Bodo, K., & Leithner, A. (2014). Adverse reactions of artificial bone graft substitutes: Lessons learned from using tricalcium phosphate geneX®. *Clinical Orthopaedics and Related Research*, 472(3), 976–982. <https://doi.org/10.1007/s11999-013-3305-z>
- Ghasemi-Mobarakeh, L., Kolahreez, D., Ramakrishna, S., & Williams, D. (2019). Key terminology in biomaterials and biocompatibility. Dalam *Current Opinion in Biomedical Engineering* (Vol. 10, hlm. 45–50). Elsevier B.V. <https://doi.org/10.1016/j.cobme.2019.02.004>
- Ghazizadeh, E., Jabbari, A. H., & Sedighi, M. (2021). In vitro corrosion-fatigue behavior of biodegradable Mg/HA composite in simulated body fluid. *Journal of Magnesium and Alloys*, 9(6), 2169–2184. <https://doi.org/10.1016/j.jma.2021.03.027>
- Godavitarne, C., Robertson, A., Peters, J., & Rogers, B. (2017). Biodegradable materials. *Orthopaedics and Trauma*, 31(5), 316–320. <https://doi.org/10.1016/j.mporth.2017.07.011>
- Goharian, A., & Abdullah, M. R. (2017). Bioinert Metals (Stainless Steel, Titanium, Cobalt Chromium). Dalam *Trauma Plating Systems* (hlm. 115–142). Elsevier. <https://doi.org/10.1016/B978-0-12-804634-0.00007-0>
- Harsimran, S., Santosh, K., & Rakesh, K. (2021). OVERVIEW OF CORROSION AND ITS CONTROL: A CRITICAL REVIEW. *Proceedings on Engineering Sciences*, 3(1), 13–24. <https://doi.org/10.24874/PES03.01.002>
- Henry, I. (2016). *Pengaruh Variasi Tegangan dan Waktu pada Anodisasi Magnesium dengan NaOH 3M terhadap Laju Korosi Magnesium*.
- Heriyanto, K., Teknologi, P., & Radioaktif, L. (2018). *KUALITAS PREPARASI SAMPEL ANALISIS KOROSI MENGGUNAKAN POTENTIOSTATE*.
- Herrera Hernández, H., M. Ruiz Reynoso, A., C. Trinidad González, J., O. González Morán, C., G. Miranda Hernández, J., Mandujano Ruiz, A., Morales Hernández, J., & Orozco Cruz, R. (2020). Electrochemical Impedance

- Spectroscopy (EIS): A Review Study of Basic Aspects of the Corrosion Mechanism Applied to Steels. Dalam *Electrochemical Impedance Spectroscopy*. IntechOpen. <https://doi.org/10.5772/intechopen.94470>
- Huzum, B., Puha, B., Necoara, R. M., Gheorghevici, S., Puha, G., Filip, A., Sirbu, P. D., & Alexa, O. (2021). Biocompatibility assessment of biomaterials used in orthopedic devices: An overview (Review). *Experimental and therapeutic medicine*, 22(5), 1315. <https://doi.org/10.3892/etm.2021.10750>
- Jaiswal, S., Kumar, R. M., Gupta, P., Kumaraswamy, M., Roy, P., & Lahiri, D. (2018). Mechanical, corrosion and biocompatibility behaviour of Mg-3Zn-HA biodegradable composites for orthopaedic fixture accessories. *Journal of the Mechanical Behavior of Biomedical Materials*, 78, 442–454. <https://doi.org/10.1016/j.jmbbm.2017.11.030>
- Johari, N. A., Alias, J., Zanurin, A., Mohamed, N. S., Alang, N. A., & Zain, M. Z. M. (2022). Anti-corrosive coatings of magnesium: A review. *Materials Today: Proceedings*, 48, 1842–1848. <https://doi.org/10.1016/j.matpr.2021.09.192>
- Jović, V. (2022). Calculation of a pure double-layer capacitance from a constant phase element in the impedance measurements. *Zastita materijala*, 63(1), 50–57. <https://doi.org/10.5937/zasmat2201050J>
- Kadhim, A., Al-Amiery, A. A., Alazawi, R., Al-Ghezi, M. K. S., & Abass, R. H. (2021). Corrosion inhibitors. A review. *International Journal of Corrosion and Scale Inhibition*, 10(1), 54–67. <https://doi.org/10.17675/2305-6894-2021-10-1-3>
- Kanazawa, M., Tsuru, K., Fukuda, N., Sakemi, Y., Nakashima, Y., & Ishikawa, K. (2017). Evaluation of carbonate apatite blocks fabricated from dicalcium phosphate dihydrate blocks for reconstruction of rabbit femoral and tibial defects. *Journal of Materials Science: Materials in Medicine*, 28(6). <https://doi.org/10.1007/s10856-017-5896-5>
- Laleh, M., Kargar, F., & Rouhaghdam, A. S. (2012). Investigation of rare earth sealing of porous micro-arc oxidation coating formed on AZ91D magnesium alloy. *Journal of Rare Earths*, 30(12), 1293–1297. [https://doi.org/10.1016/S1002-0721\(12\)60223-3](https://doi.org/10.1016/S1002-0721(12)60223-3)
- Li, L., Zhang, M., Li, Y., Zhao, J., Qin, L., & Lai, Y. (2017). Corrosion and biocompatibility improvement of magnesium-based alloys as bone implant materials: A review. Dalam *Regenerative Biomaterials* (Vol. 4, Nomor 2, hlm. 129–137). Oxford University Press. <https://doi.org/10.1093/rb/rbx004>
- Lin, C. S., & Fu, Y. C. (2006). Characterization of Anodic Films on AZ31 Magnesium Alloys in Alkaline Solutions Containing Fluoride and Phosphate

- Anions. *Journal of The Electrochemical Society*, 153(10), B417. <https://doi.org/10.1149/1.2257987>
- Lkhagvaa, T., Rehman, Z. U., & Choi, D. (2021). Post-anodization methods for improved anticorrosion properties: a review. *Journal of Coatings Technology and Research*, 18(1), 1–17. <https://doi.org/10.1007/s11998-020-00367-8>
- maarten. (2011). *Electrochemical Impedance Spectroscopy (EIS Part 1-Basic Principles)*.
- Magar, H. S., Hassan, R. Y. A., & Mulchandani, A. (2021). Electrochemical impedance spectroscopy (Eis): Principles, construction, and biosensing applications. Dalam *Sensors* (Vol. 21, Nomor 19). MDPI. <https://doi.org/10.3390/s21196578>
- Mehdizade, M., Eivani, A. R., Tabatabaei, F., Mousavi Anijdan, S. H., & Jafarian, H. R. (2023). Enhanced in-vitro biodegradation, bioactivity, and mechanical properties of Mg-based biocomposite via addition of calcium-silicate-based bioceramic through friction stir processing as resorbable temporary bone implant. *Journal of Materials Research and Technology*, 26, 4007–4023. <https://doi.org/10.1016/j.jmrt.2023.08.128>
- Meyer, Y. A., Menezes, I., Bonatti, R. S., Bortolozzo, A. D., & Osório, W. R. (2022). EIS Investigation of the Corrosion Behavior of Steel Bars Embedded into Modified Concretes with Eggshell Contents. *Metals*, 12(3), 417. <https://doi.org/10.3390/met12030417>
- Mohammed, A., & Abdullah, A. (2018). *SCANNING ELECTRON MICROSCOPY (SEM): A REVIEW*.
- Mousa, H. M., Chan, H. P., & Kim, C. S. (2017). Surface Modification of Magnesium and its Alloys Using Anodization for Orthopedic Implant Application. Dalam *Magnesium Alloys*. InTech. <https://doi.org/10.5772/66341>
- Mousa, H. M., Hussein, K. H., Woo, H. M., Park, C. H., & Kim, C. S. (2015). One-step anodization deposition of anticorrosive bioceramic compounds on AZ31B magnesium alloy for biomedical application. *Ceramics International*, 41(9), 10861–10870. <https://doi.org/10.1016/j.ceramint.2015.05.027>
- Mousa, H. M., Lee, D. H., Park, C. H., & Kim, C. S. (2015). A novel simple strategy for in situ deposition of apatite layer on AZ31B magnesium alloy for bone tissue regeneration. *Applied Surface Science*, 351, 55–65. <https://doi.org/10.1016/j.apsusc.2015.05.099>
- Mousavian, S. M. H., & Tabaian, S. H. (2022). The effect of anodizing electrolyte composition on electrochemical properties of anodized magnesium. *Anti-*

- Corrosion Methods and Materials*, 69(2), 194–203.
<https://doi.org/10.1108/ACMM-10-2021-2557>
- Niu, B., Shi, P., Shanshan, E., Wei, D., Li, Q., & Chen, Y. (2016). Preparation and characterization of HA sol–gel coating on MAO coated AZ31 alloy. *Surface and Coatings Technology*, 286, 42–48.
<https://doi.org/10.1016/j.surfcoat.2015.11.056>
- Okonkwo, P. C., Belgacem, I. Ben, Ige, O. O., Emori, W., Uzoma, P. C., Eqbal, M. O., & Bhowmik, H. (2021). Potentiodynamic polarization test as a versatile tool for bipolar plates materials at start-up and shut-down environments: a review. *International Journal of Green Energy*, 18(11), 1193–1202.
<https://doi.org/10.1080/15435075.2021.1904948>
- Oliveira, L. A., & Antunes, R. A. (2018). *Effect of the anodizing potential on the corrosion behavior of the AZ31B magnesium alloy*.
- Pedefferri, P. (2019). *Engineering Materials Corrosion Science and Engineering Edited by Luciano Lazzari and MariaPia Pedefferri*.
<http://www.springer.com/series/4288>
- Pu, Y., Hu, J., Yao, T., Li, L., Zhao, J., & Guo, Y. (2021). Influence of anodization parameters on film thickness and volume expansion of thick- and large-sized anodic aluminum oxide film. *Journal of Materials Science: Materials in Electronics*, 32(10), 13708–13718. <https://doi.org/10.1007/s10854-021-05948-w>
- Purwanto, Y., Teknologi, P., & Radioaktif, L. (2019). *Prosiding Hasil Penelitian dan Kegiatan Tahun 2018 PENGUKURAN LAJU KOROSI STAINLESS STEEL DAN BAJA KARBON DENGAN METODE TAFEL DAN POLARIZATION RESISTENCE*.
- Rahman, Z. U., Deen, K. M., & Haider, W. (2019). Controlling corrosion kinetics of magnesium alloys by electrochemical anodization and investigation of film mechanical properties. *Applied Surface Science*, 484, 906–916.
<https://doi.org/10.1016/j.apsusc.2019.02.168>
- Rahyussalim, A. J., Marsetio, A. F., Kamal, A. F., Supriadi, S., Setyadi, I., Pribadi, P. M., Mubarak, W., & Kurniawati, T. (2021). Synthesis, Structural Characterization, Degradation Rate, and Biocompatibility of Magnesium-Carbonate Apatite (Mg-Co₃Ap) Composite and Its Potential as Biodegradable Orthopaedic Implant Base Material. *Journal of Nanomaterials*, 2021.
<https://doi.org/10.1155/2021/6615614>
- Rahyussalim, A. J., Supriadi, S., Marsetio, A. F., Pribadi, P. M., & Suharno, B. (2019). The potential of carbonate apatite as an alternative bone substitute material. Dalam *Medical Journal of Indonesia* (Vol. 28, Nomor 1, hlm. 92–

- 97). Faculty of Medicine, Universitas Indonesia.
<https://doi.org/10.13181/mji.v28i1.2681>
- Reinke, S. J., Gobbi, G., Rocha, V. J., Sousa, Y. P., & Coutinho, T. P. (2020). Biomaterial: Concepts and Basics Properties. Dalam *European International Journal of Science and Technology* (Vol. 9, Nomor 2). www.eijst.org.uk
- Ribeiro, D. V., & Abrantes, J. C. C. (2016). Application of electrochemical impedance spectroscopy (EIS) to monitor the corrosion of reinforced concrete: A new approach. *Construction and Building Materials*, *111*, 98–104. <https://doi.org/10.1016/j.conbuildmat.2016.02.047>
- Salman, S. A., Mori, R., Ichino, R., & Okido, M. (2010). Effect of Anodizing Potential on the Surface Morphology and Corrosion Property of AZ31 Magnesium Alloy. *MATERIALS TRANSACTIONS*, *51*(6), 1109–1113. <https://doi.org/10.2320/matertrans.M2009380>
- Salman, S. A., & Okido, M. (2013). Anodization of magnesium (Mg) alloys to improve corrosion resistance. Dalam *Corrosion Prevention of Magnesium Alloys: A volume in Woodhead Publishing Series in Metals and Surface Engineering* (hlm. 197–231). Elsevier Ltd. <https://doi.org/10.1533/9780857098962.2.197>
- Setyadi, I., Marsetio, A. F., Kamal, A. F., Rahyussalim, Supriadi, S., & Suharno, B. (2020). Microstructure and microhardness of carbonate apatite particle-reinforced Mg composite consolidated by warm compaction for biodegradable implant application. *Materials Research Express*, *7*(5). <https://doi.org/10.1088/2053-1591/ab7d70>
- Setyadi, I., Nyoman Jujur, I., Wibisono, M., Manawan, M., Adhitya, K., Hidayat, A., Fauzi Kamal, A., Suharno, B., & Supriadi, S. (2024). Composite of Magnesium and Carbonate Apatite for Biodegradable Bone Implants: A Comparative Study on Sintering and Extrusion Techniques. *International Journal o Advanced Science Engineering Information Technology*, *14*(1).
- Setyadi, I., Pribadi, P. M., Marsetio, A. F., Kamal, A. F., Rahyusalim, Suharno, B., & Supriadi, S. (2020). Characteristics investigation of the initial development of miniplate made from composite of magnesium/carbonate apatite fabricated by powder metallurgy method for biodegradable implant applications. Dalam *Key Engineering Materials: Vol. 833 KEM* (hlm. 194–198). Trans Tech Publications Ltd. <https://doi.org/10.4028/www.scientific.net/KEM.833.194>
- Setyadi, I., Sudiro, T., Hermanto, B., Oktari, P. R., Kamal, A. F., Rahyussalim, A. J., Suharno, B., & Supriadi, S. (2022). Fabrication of Magnesium-Carbonate Apatite by Conventional Sintering and Spark Plasma Sintering for Orthopedic

- Implant Applications. *Sains Malaysiana*, 51(3), 883–894. <https://doi.org/10.17576/jsm-2022-5103-22>
- Sharma, A. K., Rani, R. U., Bhojaraj, H., & Narayanamurthy, H. (1993). Galvanic black anodizing on Mg-Li alloys. *Journal of Applied Electrochemistry*, 23(5), 500–507. <https://doi.org/10.1007/BF00707629>
- Sharma, N. R., Subburaj, K., Sandhu, K., & Sharma, V. (2021). Applications of 3D printing in Biomedical Engineering. Dalam *Applications of 3D Printing in Biomedical Engineering*. Springer Singapore. <https://doi.org/10.1007/978-981-33-6888-0>
- Sheikh, Z., Najeeb, S., Khurshid, Z., Verma, V., Rashid, H., & Glogauer, M. (2015). Biodegradable Materials for Bone Repair and Tissue Engineering Applications. *Materials*, 8(9), 5744–5794. <https://doi.org/10.3390/ma8095273>
- Su, J. long, Teng, J., Xu, Z. li, & Li, Y. (2020). Biodegradable magnesium-matrix composites: A review. Dalam *International Journal of Minerals, Metallurgy and Materials* (Vol. 27, Nomor 6, hlm. 724–744). University of Science and Technology Beijing. <https://doi.org/10.1007/s12613-020-1987-2>
- Thomas, S., Medhekar, N. V., Frankel, G. S., & Birbilis, N. (2015). Corrosion mechanism and hydrogen evolution on Mg. Dalam *Current Opinion in Solid State and Materials Science* (Vol. 19, Nomor 2, hlm. 85–94). Elsevier Ltd. <https://doi.org/10.1016/j.cossms.2014.09.005>
- Ul-Hamid, A. (2018). *A Beginners' Guide to Scanning Electron Microscopy*. Springer International Publishing. <https://doi.org/10.1007/978-3-319-98482-7>
- Verma, C., Olasunkanmi, L. O., Bahadur, I., Lgaz, H., Quraishi, M. A., Haque, J., Sherif, E.-S. M., & Ebenso, E. E. (2019). Experimental, density functional theory and molecular dynamics supported adsorption behavior of environmental benign imidazolium based ionic liquids on mild steel surface in acidic medium. *Journal of Molecular Liquids*, 273, 1–15. <https://doi.org/10.1016/j.molliq.2018.09.139>
- Verma, C., Olasunkanmi, L. O., Ebenso, E. E., Quraishi, M. A., & Obot, I. B. (2016). Adsorption Behavior of Glucosamine-Based, Pyrimidine-Fused Heterocycles as Green Corrosion Inhibitors for Mild Steel: Experimental and Theoretical Studies. *The Journal of Physical Chemistry C*, 120(21), 11598–11611. <https://doi.org/10.1021/acs.jpcc.6b04429>
- Wang, S., Zhang, J., Gharbi, O., Vivier, V., Gao, M., & Orazem, M. E. (2021). Electrochemical impedance spectroscopy. *Nature Reviews Methods Primers*, 1(1), 41. <https://doi.org/10.1038/s43586-021-00039-w>

- Wierzbicka, E., Vaghefinazari, B., Mohedano, M., Visser, P., Posner, R., Blawert, C., Zheludkevich, M., Lamaka, S., Matykina, E., & Arrabal, R. (2022). Chromate-Free Corrosion Protection Strategies for Magnesium Alloys—A Review: Part II—PEO and Anodizing. Dalam *Materials* (Vol. 15, Nomor 23). MDPI. <https://doi.org/10.3390/ma15238515>
- Wingender, B., Azuma, M., Krywka, C., Zaslansky, P., Boyle, J., & Deymier, A. (2021). Carbonate substitution significantly affects the structure and mechanics of carbonated apatites. *Acta Biomaterialia*, *122*, 377–386. <https://doi.org/10.1016/j.actbio.2021.01.002>
- Xue, D., Yun, Y., Schulz, M. J., & Shanov, V. (2011). Corrosion protection of biodegradable magnesium implants using anodization. *Materials Science and Engineering: C*, *31*(2), 215–223. <https://doi.org/10.1016/j.msec.2010.08.019>
- Yang, Y., He, C., Dianyu E, Yang, W., Qi, F., Xie, D., Shen, L., Peng, S., & Shuai, C. (2020). Mg bone implant: Features, developments and perspectives. Dalam *Materials and Design* (Vol. 185). Elsevier Ltd. <https://doi.org/10.1016/j.matdes.2019.108259>
- Yu, W., Sun, R., Guo, Z., Wang, Z., He, Y., Lu, G., Chen, P., & Chen, K. (2019). Novel fluoridated hydroxyapatite/MAO composite coating on AZ31B magnesium alloy for biomedical application. *Applied Surface Science*, *464*, 708–715. <https://doi.org/10.1016/j.apsusc.2018.09.148>
- Yu, Z., Xu, X., Shi, K., Du, B., Han, X., Xiao, T., Li, S., Liu, K., & Du, W. (2023). Development and characteristics of a low rare-earth containing magnesium alloy with high strength-ductility synergy. *Journal of Magnesium and Alloys*, *11*(5), 1629–1642. <https://doi.org/10.1016/j.jma.2022.01.005>
- Zaffora, A., Di Franco, F., Virtù, D., Carfi Pavia, F., Gherzi, G., Virtanen, S., & Santamaria, M. (2021). Tuning of the Mg Alloy AZ31 Anodizing Process for Biodegradable Implants. *ACS Applied Materials and Interfaces*, *13*(11), 12866–12876. <https://doi.org/10.1021/acsami.0c22933>
- Zhang, G., Wu, L., Tang, A., Ma, Y., Song, G.-L., Zheng, D., Jiang, B., Atrens, A., & Pan, F. (2018). Active corrosion protection by a smart coating based on a MgAl-layered double hydroxide on a cerium-modified plasma electrolytic oxidation coating on Mg alloy AZ31. *Corrosion Science*, *139*, 370–382. <https://doi.org/10.1016/j.corsci.2018.05.010>
- Zhang, Y., Yan, C., Wang, F., Lou, H., & Cao, C. (2002). Study on the environmentally friendly anodizing of AZ91D magnesium alloy. *Surface and Coatings Technology*, *161*(1), 36–43. [https://doi.org/10.1016/S0257-8972\(02\)00342-0](https://doi.org/10.1016/S0257-8972(02)00342-0)

- Zhou, H., Liang, B., Jiang, H., Deng, Z., & Yu, K. (2021). Magnesium-based biomaterials as emerging agents for bone repair and regeneration: from mechanism to application. Dalam *Journal of Magnesium and Alloys* (Vol. 9, Nomor 3, hlm. 779–804). National Engg. Reaserch Center for Magnesium Alloys. <https://doi.org/10.1016/j.jma.2021.03.004>
- Zoroddu, M. A., Aaseth, J., Crisponi, G., Medici, S., Peana, M., & Nurchi, V. M. (2019). The essential metals for humans: a brief overview. Dalam *Journal of Inorganic Biochemistry* (Vol. 195, hlm. 120–129). Elsevier Inc. <https://doi.org/10.1016/j.jinorgbio.2019.03.013>