

**ANALISIS PENGARUH PENAMBAHAN LAPISAN rGO PADA LaFeO₃
DENGAN DOPING Pd MENGGUNAKAN *DENSITY FUNCTIONAL
THEORY UNTUK SENSOR GAS***

SKRIPSI

diajukan untuk memenuhi salah satu syarat untuk memperoleh gelar Sarjana Sains
Program Studi Fisika Kelompok Bidang Kajian Fisika Material



Oleh

Rumaisya Az-zahra

2007015

PROGRAM STUDI FISIKA

**FAKULTAS PENDIDIKAN MATEMATIKA DAN ILMU PENGETAHUAN ALAM
UNIVERSITAS PENDIDIKAN INDONESIA**

2024

**ANALISIS PENGARUH PENAMBAHAN LAPISAN rGO PADA LaFeO₃
DENGAN DOPING Pd MENGGUNAKAN *DENSITY FUNCTIONAL THEORY*
UNTUK SENSOR GAS**

Oleh
Rumaisya Az-zahra

Diajukan untuk memenuhi sebagian syarat dalam memperoleh gelar Sarjana Sains

Program Studi Fisika
Konsentrasi Fisika Material
FPMIPA UPI

© Rumaisya Az-zahra

Universitas Pendidikan Indonesia

Juli 2024

Hak cipta dilindungi Undang-Undang.

Skripsi ini tidak boleh diperbanyak seluruhnya atau sebagian, dengan dicetak
ulang, difotocopy atau cara lainnya tanpa izin dari penulis.

LEMBAR PENGESAHAN

RUMAISYA AZ-ZAHRA

ANALISIS PENGARUH PENAMBAHAN LAPISAN rGO PADA LaFeO₃ DENGAN DOPING Pd MENGGUNAKAN DENSITY FUNCTIONAL THEORY UNTUK SENSOR GAS

disetujui dan disahkan oleh :

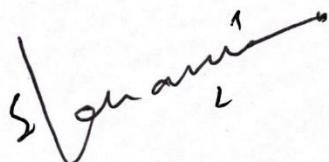
Pembimbing I



Prof. Dr. Endi Suhendi, M.Si.

NIP. 197905012003121001

Pembimbing II

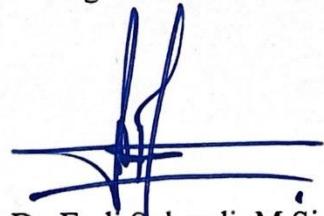


Dr. Selly Feranie, M.Si.

NIP. 197411081999032004

Mengetahui,

Ketua Program Studi Fisika



Prof. Dr. Endi Suhendi, M.Si.

NIP. 197905012003121001

PERNYATAAN

Dengan ini saya menyatakan bahwa skripsi dengan judul “ANALISIS PENGARUH PENAMBAHAN LAPISAN rGO PADA LaFeO₃ DENGAN DOPING Pd MENGGUNAKAN *DENSITY FUNCTIONAL THEORY* UNTUK SENSOR GAS” ini beserta seluruh isinya adalah benar-benar karya sendiri. Saya tidak melakukan penjiplakan atau pengutipan dengan cara-cara yang tidak sesuai dengan etika ilmu yang berlaku dalam masyarakat keilmuan. Atas pernyataan ini, saya siap menanggung risiko/sanksi apabila di kemudian hari ditemukan adanya pelanggaran etika keilmuan atau ada klaim dari pihak lain terhadap keaslian karya saya ini.

Bandung, Juli 2024

Yang membuat pernyataan

Rumaisya Az-zahra

NIM. 2007015

KATA PENGANTAR

Segala puji dan rasa syukur dipanjangkan kepada Allah SWT atas segala rahmat dan karunia-Nya, skripsi dengan judul “Analisis Pengaruh Penambahan Lapisan rGO Pada LaFeO₃ Dengan Doping Pd Menggunakan *Density Functional Theory* Untuk Sensor Gas” dapat diselesaikan dengan baik.

Skripsi ini telah disusun sebagai syarat untuk memenuhi mencapai gelar Sarjana Sains Fisika (S1) dari program studi Fisika, Fakultas Pendidikan Matematika dan Ilmu Pengetahuan Alam, Universitas Pendidikan Indonesia. Penulis menyadari bahwa skripsi ini masih memiliki kekurangan dan jauh dari sempurna. Oleh karena itu, kritik dan saran dari berbagai pihak akan sangat membangun untuk penulisan skripsi ini. Semoga skripsi ini dapat bermanfaat bagi para pembaca dan mendorong penelitian selanjutnya. Akhir kata, penulis mengucapkan terima kasih.

Bandung, Juli 2024

Rumaisya Az-zahra

NIM. 2007015

UCAPAN TERIMA KASIH

Penulisan skripsi tidak terlepas dari banyaknya dukungan serta bantuan dari banyak pihak. Tidak terlupa juga, puji dan syukur kepada Allah SWT atas segala karunia-Nya. Penulis juga mengucapkan terima kasih yang sebesar-besarnya kepada :

1. Bapak Prof. Dr. Endi Suhendi, M.Si., selaku Dosen Pembimbing I dan Ketua Program Studi Fisika yang telah memberikan bimbingan, masukan, kritik, saran, dukungan, serta semangat sejak dimulainya penelitian hingga akhir penulisan skripsi.
2. Ibu Dr. Selly Feranie, M.Si., selaku Dosen Pembimbing II yang telah memberikan masukan, kritik, dan saran untuk membangun penulisan skripsi sejak dimulainya penelitian hingga akhir penulisan skripsi.
3. Ibu Prof. Dr. Lilik Hasanah, M.Si., selaku Dosen Pembimbing Akademik yang selalu mengarahkan dan membimbing dari awal perkuliahan hingga selesai perkuliahan.
4. Seluruh dosen Fisika yang telah mengajar dan membimbing selama perkuliahan.
5. Badan Riset dan Inovasi Nasional (BRIN), selaku penyedia fasilitas layanan *High Performance Computing* selama proses komputasi.
6. Kak Maulana Ibrohim S.Si dan kak Robi Dwiputra Satria S.Si, sebagai pembimbing dan memberikan arahan selama proses penggerjaan komputasi.
7. Kedua orang tua yang luar biasa memberikan dukungan, semangat, membiayai seluruh pendidikan yang telah ditempuh, kasih sayang, dan motivasi yang selalu diberikan untuk mengerjakan dan menyelesaikan skripsi ini.
8. Keluarga besar yang senantiasa memberikan dukungan, motivasi, dan doa selama penggerjaan dan penulisan skripsi.
9. Teman satu riset sensor gas yang selalu memberikan dukungan dan masukan satu sama lain serta teman-teman Fisika C 2020 yang telah membersamai proses perkuliahan.
10. Sahabat-sahabat terdekat yang senantiasa memberikan banyak dukungan dan semangat selama penggerjaan skripsi.

ABSTRAK

Etanol merupakan salah satu jenis alkohol yang mudah menguap dan terdapat dampak negatif bagi kesehatan dan lingkungan sehingga diperlukan pengembangan komponen penginderaan gas etanol yaitu sensor gas. LaFeO₃ merupakan material perovskite ABO₃ yang banyak diaplikasikan pada sensor gas karena komposisi materialnya dan stabilitas parameter penginderaannya. Namun, LaFeO₃ masih terbatas sifat penginderaan gasnya sehingga diperlukan *doping* dalam peningkatan kinerja sensor gas yaitu Palladium. Selain itu, ditambahkan juga lapisan rGO untuk memperkuat energi adsorpsi. Pada penelitian ini dilakukan proses *Density Functional Theory* (DFT) dengan metode SCF, vc-relax, *bands*, serta *post-processing bands*. Hasil dari energi adsorpsi molekul gas etanol pada Pd-LaFeO₃ -2,01 eV dan Pd-LaFeO₃@rGO -2,29 eV. Penambahan rGO diketahui meningkatkan energi adsorpsi (negatif). Kemudian, untuk nilai *band gap* Pd-LaFeO₃ sebelum terpapar gas 2,34 eV dan saat terpapar 2,06 eV. Kemudian, Pd-LaFeO₃@rGO sebelum terpapar sekitar 0,11 eV dan saat terpapar 0,05 eV. Penyempitan struktur pita energi membuat nilai energi celah pita mengecil akibat terciptanya banyak *hole*. Dari hasil perubahan antara energi adsorpsi dan celah pita diketahui bahwa penambahan rGO pada Pd-LaFeO₃ potensial untuk dijadikan kandidat aplikasi sensor gas dari sisi kuatnya adsorpsi dan rendahnya *band gap*.

Kata Kunci : Sensor gas, LaFeO₃, Palladium, *reduce graphene oxide*, *density functional theory*

ABSTRACT

Ethanol is a type of volatile alcohol that poses negative effects on health and the environment, necessitating the development of ethanol gas sensing components, namely gas sensors. LaFeO₃ is a perovskite ABO₃ material commonly used in gas sensors due to its material composition and stable sensing parameters. However, LaFeO₃ has limited gas sensing capabilities, requiring enhancement through doping with Palladium to improve sensor performance. Additionally, an rGO layer is added to strengthen adsorption energy. This study employed Density Functional Theory (DFT) using SCF, vc-relax, bands, and post-processing bands methods. The results showed ethanol gas molecule adsorption energy on Pd-LaFeO₃ was -2,01 eV and on Pd-LaFeO₃@rGO was -2,29 eV. The addition of rGO is known to increase (negative) adsorption energy. Furthermore, the band gap of Pd-LaFeO₃ before gas exposure was 2,34 eV and 2,06 eV when exposed. For Pd-LaFeO₃@rGO, the band gap was around 0,11 eV before exposure and 0,05 eV when exposed. The narrowing of the energy band structure reduced the band gap value due to the creation of more holes. The results indicate that adding rGO to Pd-LaFeO₃ is a promising candidate for gas sensor applications, given its strong adsorption and low band gap.

Keywords: Gas sensor, LaFeO₃, Palladium, reduced graphene oxide, density functional theory

DAFTAR ISI

PERNYATAAN	ii
KATA PENGANTAR	iv
UCAPAN TERIMA KASIH.....	v
ABSTRAK	vi
ABSTRACT	vii
DAFTAR ISI.....	viii
DAFTAR GAMBAR	xi
DAFTAR LAMPIRAN.....	xiii
BAB I PENDAHULUAN	1
1.1 Latar Belakang	1
1.2 Rumusan Masalah.....	4
1.3 Batasan Masalah Penelitian.....	5
1.4 Tujuan Penelitian.....	5
1.5 Manfaat Penelitian.....	5
1.6 Struktur Organisasi Skripsi	5
BAB II KAJIAN PUSTAKA	7
2.1 Sensor Gas.....	7
2.2 Etanol	8
2.3 Material Perovskite LaFeO ₃ sebagai Sensor Gas	10
2.4 Doping Pd.....	11
2.5 Lapisan rGO	12
2.6 Energi Adsorpsi	13
2.7 Mekanisme Adsorpsi	14
2.8 Energi Celah Pita (<i>band gap</i>)	15

2.9	Density Functional Theory (DFT)	16
2.9.1	Generalized-Gradient Approximation (GGA)	17
2.9.2	GGA+U	17
2.9.3	Potensial Semu	18
2.9.4	<i>Self-Consistent Field</i> (SCF)	18
2.9.5	Variabel Cell Relaxation	19
BAB III METODE PENELITIAN		20
3.1	Waktu dan Tempat Penelitian	20
3.2	Spesifikasi perangkat keras dan perangkat lunak	20
3.3	Prosedur Penelitian	20
3.4	Diagram Alir Penelitian	21
3.5	Tahapan Komputasi	22
3.5.1	Menyiapkan Struktur Pd- LaFeO ₃	22
3.5.2	Menyiapkan rGO	22
3.5.3	Menyiapkan Struktur Gas Etanol	23
3.5.4	Menentukan Konvergensi Struktur dengan Kalkulasi SCF	23
3.5.5	Menentukan Jarak Optimal Molekul Gas Etanol	24
3.5.6	Menentukan Jarak Optimal rGO	24
3.5.7	Menentukan Energi Adsorpsi	25
3.5.8	Menetukan Celah Pita (Band Gap)	25
BAB IV HASIL DAN PEMBAHASAN		27
4.1	Konvergensi Struktur dengan Kalkulasi SCF	27
4.1.1	Konvergensi harga <i>k-points</i>	27
4.1.2	Konvergensi nilai energi <i>cut-off</i>	28
4.2	Jarak Optimal Molekul Gas Etanol	30
4.3	Jarak Optimal rGO	31

4.4	Energi Adsorpsi	32
4.4.1	Energi Adsorpsi Molekul Etanol pada Pd- LaFeO ₃	33
4.4.2	Energi Adsorpsi Molekul Etanol pada Pd- LaFeO ₃ @rGO	33
4.5	Analisis Hasil Energi Adsorpsi	34
4.6	Energi Celah Pita (<i>Band Gap</i>).....	36
4.6.1	<i>Band Gap</i> Pd-LaFeO ₃ sebelum dan saat terpapar gas etanol	36
4.6.2	<i>Band Gap</i> Pd-LaFeO ₃ @rGO sebelum dan saat terpapar gas etanol..	37
4.7	Analisis Hasil Struktur Pita dan Energi Celah Pita (<i>Band Gap</i>)	38
BAB V SIMPULAN, IMPLIKASI, DAN REKOMENDASI		42
5.1	Simpulan	42
5.2	Implikasi.....	42
5.3	Rekomendasi	42
DAFTAR PUSTAKA		44
LAMPIRAN		59

DAFTAR GAMBAR

Gambar 2. 1 Contoh Diagram skematik Mekanisme material SnO ₂ NPs/rGO ketika berada di udara (air) dan di etanol (Gai dkk., 2022)	8
Gambar 2. 2 Ilustrasi Struktur Etanol	9
Gambar 2. 3 Ilustrasi adsorption physisorption dan chemisorption (Kennedy dkk., 2018)	14
Gambar 3. 1 Diagram Alir Penelitian	21
Gambar 3. 2 Ilustrasi grafis struktur (a) LaFeO ₃ (b) Pd-LaFeO ₃	22
Gambar 3. 3 Ilustrasi grafis struktur rGO	23
Gambar 3. 4 Koordinat dan Cell Parameter yang dihasilkan dari Vc-relax.....	26
Gambar 4. 1 Plot Konvergensi harga k-points	28
Gambar 4. 2 Plot Konvergensi nilai energi cut-off.....	29
Gambar 4. 3 Ilustrasi Grafik Struktur (a) Jarak Pd- LaFeO ₃ dengan molekul etanol, (b) Jarak Pd- LaFeO ₃ dengan rGO	30
Gambar 4. 4 Plot variasi jarak molekul etanol terhadap energi total.....	31
Gambar 4. 5 Plot Variasi Jarak rGO terhadap Energi Total	32
Gambar 4. 6 Ilustrasi Grafis Struktur (a) Adsorben Pd-LaFeO ₃ dengan adsorbat molekul etanol, (b) Adsorben Pd-LaFeO ₃ @ rGO dengan Adsorbat Molekul Etanol	33
Gambar 4. 7 Struktur bands Pd-LaFeO ₃ sebelum terpapar etanol	36
Gambar 4. 8 Struktur Bands Pd-LaFeO ₃ saat terpapar etanol	37
Gambar 4. 9 Struktur bands Pd-LaFeO ₃ @rGO sebelum terpapar etanol	38
Gambar 4. 10 Struktur bands Pd-LaFeO ₃ @rGO saat terpapar etanol.....	38

DAFTAR TABEL

Tabel 4. 1 Energi adsorpsi material terhadap molekul gas tertentu	34
Tabel 4. 2 Hasil Nilai Band Gap Pd-LaFeO ₃ tanpa dan dengan rGO.....	39

DAFTAR LAMPIRAN

Lampiran 1 Input File Quantum Espresso untuk menentukan konvergensi kpoints	59
Lampiran 2 Input File Quantum Espresso untuk menentukan konvergensi energi cut-off.....	61
Lampiran 3 Input File Quantum Espresso menentukan Jarak optimal molekul etanol.....	63
Lampiran 4 Input File Quantum Espresso untuk menentukan jarak optimal lapisan reduce graphene oxide	65
Lampiran 5 Input File Quantum Espresso untuk menentukan energi total adsorben Pd-LaFeO ₃	68
Lampiran 6 Input File Quantum Espresso untuk menentukan energi total adsorbat etanol.....	70
Lampiran 7 Input File Quantum Espresso untuk menentukan energi sistem Pd-LaFeO ₃ dan etanol.....	72
Lampiran 8 Input File Quantum Espresso untuk menentukan energi total adsorben Pd-LaFeO ₃ @rGO.....	74
Lampiran 9 Input File Quantum Espresso untuk menentukan energi total sistem Pd-LaFeO ₃ @rGO.....	77
Lampiran 10 Input File Quantum Espresso untuk menentukan SCF Pd-LaFeO ₃ .80	
Lampiran 11 Input File Quantum Espresso untuk menentukan bands Pd-LaFeO ₃ 82	
Lampiran 12 Input File Quantum Espresso untuk post-processing bands Pd-LaFeO ₃	84

DAFTAR PUSTAKA

- Agboola, O. D., & Benson, N. U. (2021). *Physisorption and Chemisorption Mechanisms Influencing Micro (Nano) Plastics-Organic Chemical Contaminants Interactions : A Review.* 9(May), 1–27. <https://doi.org/10.3389/fenvs.2021.678574>
- Agusriyadin. (2020). *Karakterisasi , Kinetika , dan Isoterm Adsorpsi Limbah Ampas.* 6(2), 104–115. <https://doi.org/10.31605/saintifik.v6i2.265>
- Akter, T., Barile, C., & Ahammad, A. J. S. (2022). *Chapter 23-Introduction and overview of carbon nanomaterial-based sensors for sustainable response*
- Alahmed, Z., Albrithen, H. A., Al-Ghamdi, A. A., & Yakuphanoglu, F. Optical band gap controlling of nanostructure Mn doped CdO thin films prepared by sol-gel spin coating method. *Optik,* 126 (5), 575-577. <https://doi.org/10.1016/j.ijleo.2015.01.005>
- Aleksanyan, M., Sayunts, A., Shahkhatuni, G., Simonyan, Z., & Shahnazaryan, G. (2022). *Gas Sensor Based on ZnO Nanostructured Film for the Detection of Ethanol Vapor.* 10(245). <https://doi.org/https://doi.org/10.3390/chemosensors10070245>
- Alharbi, A. A., Weimar, U., & Bârsan, N. (2019). *Gas Sensing Mechanism Investigation of LaFeO₃ Perovskite-Type Oxides via Operando Technique* †. 14–16. <https://doi.org/10.3390/proceedings2019014051>
- Ali, S., Gupta, A., & Shafiei, M. (2021). *Recent Advances in Perylene Diimide-Based Active Materials in Electrical Mode Gas Sensing.* 9(30), 1–32. <https://doi.org/10.3390/chemosensors9020030>
- Araujo, P. De, Paiva, M. P., Moises, L. A., Santo, G. S., Blanco, K. C., Chiquito, A. J., & Amorim, C. A. (2023). *Improving Hazardous Gas Detection Behavior with Palladium Decorated SnO₂ Nanobelts Networks.*
- Bahamon, D., Khalil, M., Belabbes, A., & Alwahedi, Y. (2021). *A DFT study of the adsorption energy and electronic interactions of the SO₂ molecule on a CoP hydrotreating catalyst* †. 2947–2957.

<https://doi.org/10.1039/c9ra10634k>

Balram, D., Lian, K., Sebastian, N., Kumar, V., Yadav, K., Patel, A., & Singh, K. (2024). Graphene-metal sulfide composite based gas sensors for environmental sustainability : A review. *Sensors International*, 5(October 2023), 100269. <https://doi.org/10.1016/j.sintl.2023.100269>

Basahel, S. N., Medkhali, A. H. A., Mokhtar, M., & Narasimharao. (2022). *Noble metal (Pd, Pt and Rh) incorporated LaFeO₃ perovskite oxides for catalytic oxidative cracking of n-propane*. <https://doi.org/10.1016/j.cattod.2021.11.032>

Bhikuning, A., Setiawan, B., Jati, J. F., Wijaya, J. D. I., & Hafnan, M. (2023). *Investigation of fuel properties and structural-functional group analysis in blending low and high boiling point fuels: the case of ethanol with fuel*. 21(3), 272–277.

Bylaska, E. J. (2017). Chapter Five – Plane-Wave DFT Methods for Chemistry. *Annual Reports in Computational Chemistry*, 13, 185-228. <https://doi.org/10.1016/bs.arcc.2017.06.006>

Callister, W. D. & Rethwisch, D. G. (1997). Materials Science And Engineering.

Chakraborty, A., Nuthalapati, S., Nag, A., & Afsarimanesh, N. (2022). *A Critical Review of the Use of Graphene-Based Gas Sensors*.

Chatterjee, S.G., Chatterjee, S., Ray, A.K., & Chakraborty, A.K. (2015). *Graphene-metal oxide nanohybrids for toxic gas sensor:A review*. <https://doi.org/10.1016/j.snb.2015.07.070>

Chen, N., Deng, D., Li, Y., Liu, X., Xing, X., & Xiao, X. (2017). TiO₂ nanoparticles functionalized by Pd nanoparticles for gas-sensing application with enhanced butane response performances. *Scientific Reports*, July, 1–11. <https://doi.org/10.1038/s41598-017-08074-y>

Chumakova, V., Marikutsa, A., Platonov, V., Khmelevsky, N., & Rumyantseva, M. (2023). Distinct Roles of Additives in the Improved Sensitivity to CO of Ag- and Pd-Modified Nanosized LaFeO₃. *Chemosensors*, 11(60). <https://www.mdpi.com/2227-9040/11/1/60#>

- Dayekh, M. L., Hussain, S. A. (2022). *Gas Sensor and Sensitivity*. IntechOpen.
doi: 10.5772/intechopen.108040
- Dhariwal, N., Yadav, P., Sanger, A., Kang, S. B., Goyat, M. S., Mishra, Y. K., & Kumar, V. (2024). Materials Advances Fabrication of a room-temperature NO₂ gas sensor with high performance at the ppb level. *Mater. Adv.*, 5(2), 4187–4199. <https://doi.org/10.1039/d4ma00168k>
- Dey, A. (2018). *Semiconductor metal oxide gas sensors : A review*
- Doubi, Y., Hartiti, B., Sidadat, M., dkk. (2022). *Theoretical validation of the properties of SnO₂ nanostructure grown by Robust Spray Pyrolysis technique for formaldehyde gas sensor.*
- Du, H., Xie, G., Su, Y., Tai, H., Du, X., & Yu, H. (2019). *A New Model and Its Application for the Dynamic Response of RGO Resistive Gas Sensor*. 1906, 1–11. <https://doi.org/10.3390/s19040889>
- Duan, K., Li, W., Zhu, C., Li, J., Xu, J., & Wang, X. (2022). Results in Physics Promoting sensitivity and selectivity of NO₂ gas sensor based on (P, N) -doped single-layer WSe₂ : A first principles study. *Results in Physics*, 34(2), 105296. <https://doi.org/10.1016/j.rinp.2022.105296>
- Erduran, V., Bekmezci, M., Bayat, R., & Sen, F. *Functionalized carbon material-based electrochemical sensors for day-to-day applications*. <https://doi.org/10.1016/B978-0-12-823788-5.00017-X>
- Fan, J., Ran, X., Fu, H., Sun, J., An, X., & Yang, X. (2024). The roles of oxygen vacancies in LaFeO₃ photocathodes in photoelectrochemical water splitting. *Applied Surface Science*, 649. <https://doi.org/10.1016/j.apsusc.2023.159011>
- Feng, Z., Lin, L., Su, L., Pang, D., & Shi, P. (2023). *Mechanism of adsorption of hazardous gases by MoTe₂ monolayers modified with nanoclusters*. <https://doi.org/10.1016/j.jece.2023.110486>
- Fu, L., You, S., Li, G., Li, X., & Fan, Z. (2023). Application of Semiconductor Metal Oxide in Chemiresistive Methane Gas Sensor : Recent Developments and Future Perspectives. *Molecules*, 28(6710). <https://www.mdpi.com/1420->

3049/28/18/6710#

- Gai, LY., Lai, RP., Dong, XH. *et al.* Recent advances in ethanol gas sensors based on metal oxide semiconductor heterojunctions. *Rare Met.* 41, 1818–1842 (2022). <https://doi.org/10.1007/s12598-021-01937-4>
- Gao, R., Gao, L., Zhang, X., Gao, S., Xu, Y., Cheng, X., Guo, G., Ye, Q., Zhou, X., Major, Z., & Huo, L. (2021). The controllable assembly of the heterojunction interface of the ZnO@rGO for enhancing the sensing performance of NO₂ at room temperature and sensing mechanism. *Sensors and Actuators, B: Chemical*, 342(February), 1–11. <https://doi.org/10.1016/j.snb.2021.130073>
- Gao, R., Yong, Y., Yuan, X., Hu, S., Hou, Q., & Kuang, Y. (2022). *First-Principles Investigation of Adsorption Behaviors and Electronic , Optical , and Gas-Sensing Properties of Pure and Pd- Decorated GeS 2 Monolayers*. <https://doi.org/10.1021/acsomega.2c05142>
- Gao, Y., Wang, X., Zhang, Q., Wang, H., Xu, G., & Wang, X. Influence of La doping on the ethanol gas sensing properties of CdSnO₃ micro-cubes. *Sensors and Actuators B: Chemical*, 394. <https://doi.org/10.1016/j.snb.2023.134447>
- Giannozzi, P., et al. (2009) Quantum Espresso: A Modular and Open-Source Software Project for Quantum Simulations of Materials. *Journal of Physics: Condensed Matter*, 21, Article ID: 395502. <https://doi.org/10.1088/0953-8984/21/39/395502>
- Gu, J., Zhang, B., Li, Y., Xu, X., Sun, G., Cao, J., & Wang, Y. (2021). *Synthesis of spindle-like Co-doped LaFeO₃ Porous microstructure for high performance n-butanol sensor*. *Sensors and Actuators B: Chemical*. <https://doi.org/10.1016/j.snb.2021.130125>
- Gordon, A. F. M., Jimenez, P. E. S., Blazquez, J. S., Perejon, A., & Maqueda, L. A. P. (2023). *Orthoferrite LaFeO₃ Ceramic Prepared by Reaction Flash Sintering*.

- Hakim, A. R., & Saputri, R. (2020). *NARRATIVE REVIEW : OPTIMASI ETANOL SEBAGAI PELARUT SENYAWA FLAVONOID DAN FENOLIK* Narrative Review : Optimization of Ethanol as a Solvent for Flavonoids and Phenolic Compounds Abstrak.
- Harun, K., Azmira, N., Deghfel, B., Kamil, M., & Azmin, A. (2020). Results in Physics DFT + U calculations for electronic , structural , and optical properties of ZnO wurtzite structure: A review. *Results in Physics*, 16(November 2019), 102829. <https://doi.org/10.1016/j.rinp.2019.102829>
- Heinrich, P., Hanslik, L., Kammer, N., & Braunbeck, P. (2020). *The Tox is in the detail : technical fundamentals for designing, performing, and interpreting, experiments on toxicity of microplastics and associated substance.*
- Heryanto, Ardiansyah, A., Rahmat, R., & Tahir, D. (2024). Science Mapping Analysis of Density Functional Theory (DFT) for Material Design: A Review. *JOM*. Doi : <https://doi.org/10.1007/s11837-024-06644-w>
- Huang, C., Zhu, Y., & Man, X. (2021). Block copolymer thin films. *Physics Reports*, 932, 1–36. <https://doi.org/10.1016/j.physrep.2021.07.005>
- Irzon, R., & Kurnia. (2019). *SKEMA FIRE ASSAY DAN ICP-MS PADA PENGUKURAN KADAR PALADIUM DALAM SAMPEL BATUAN FIRE ASSAY AND ICP-MS SCHEMES ON THE PALLADIUM.* 187–194.
- Ishikawa, T., Hayakawa, D., dkk. (2015). *Ab initio studies of the structure of and atomic interactions in cellulose III crystals.*
- Jana, A., Bergsman, D. S., & Grossman, C. (2021). *Nanoscale Advances Adsorption-based membranes for air separation using transition metal oxides* †. 4502–4512. <https://doi.org/10.1039/D1NA00307K>
- Ji, H., Zeng, W., & Li, Y. (2019). *Gas Sensing mechanism of metal oxide semiconductor : a focus review.*
- Jia, L., Chen, J., Cui, X., Wang, Z., Zeng, W., & Zhou, Q. (2022). *Gas Sensing Mechanism and Adsorption Properties of C₂H₄ and CO Molecules on the Ag₃ – HfSe₂ Monolayer : A First-Principle Study.* 10(May), 1–11.

<https://doi.org/10.3389/fchem.2022.911170>

Jian, J., Chang, Y., Chang, S., & Chang, S.-J. (2021). *High Response of Ethanol Gas Sensor Based on NiO-Doped Apple Pectin by the Solution Process.* Journals Coatings 11(9). Doi : <https://doi.org/10.3390/coatings11091073>

Kaewpanha, M., Suriwong, T., Wamae, W., & Nunocha, P. (2019). Synthesis and Characterization of Sr-doped LaFeO₃ perovskite by sol-gel auto-combustion method. *IOP Conf. Series: Journal of Physics: Conf,* 1259. <https://doi.org/10.1088/1742-6596/1259/1/012017>

Kang, S., Sanger, A., Jeong, M. H., Baik, J. M., & Choi, K. J. (2023). Heterogeneous Stacking of Reduced Graphene Oxide on ZnO nanowires for NO₂ gas sensors with dramatically improved response and high sensitivity. *Sensors and Actuator B: Chemical,* 379, 133196. Doi : <https://doi.org/10.1016/j.snb.2022.133196>

Kang, Y., Zhang, L., Wang, W., & Yu, F. (2021). Ethanol Sensing Properties and First Principles Study of Au Supported on Mesoporous ZnO Derived from Metal Organic. *Sensors,* 21(4352). <https://doi.org/https://doi.org/10.3390/s21134352>

Kennedy, K., Maseka, K., & Mbulo, M. Selected Adsorbents for Removal of Contaminants from Wastewater: Towards Engineering Clay Minerals. *Journal of Applied Science* 08(08):355-369. Doi : <http://dx.doi.org/10.4236/ojapps.2018.88027>

Khalifeh, S. (2020). *Optimization of Electrical Electronic and Optical Properties of Organic Electronic Structures.* <https://doi.org/10.1016/B978-1-927885-67-3.50009-2>

Klenam, D., Rahbar, N., & Soboyejo, W. (2023). *Critical Revie of Factors Hindering Scalability of Complex Concentrated Alloy.*

Ko, Y., Tsai, H., Lin, K., Chen, Y., & Yang, H. (2017). Reusable macroporous photonic crystal-based ethanol vapor detectors by doctor blade coating. *Journal of Colloid And Interface Science,* 487, 360–369.

- <https://doi.org/10.1016/j.jcis.2016.10.061>
- Kou, L., Frauenheim, T., & Changfeng, C. (2014). *Phosphorene as a Superior Gas Sensor : Selective Adsorption and Distinct I-V Response.* <https://doi.org/10.1021/jz501188k>
- Kuchi, P. S., Roshan, H., Sheikhi, M. H. (2020). *A novel room temperature ethanol sensor based on PbS : SnS₂ nanocomposite with enhanced ethanol sensing properties.* <https://doi.org/10.1016/j.jallcom.2019.152666>
- Kumar, A., Zhao, Y., Mohammadi, M. M., Liu, J., Thundat, T., & Swihart, M. T. (2022). *Palladium Nanosheet-Based Dual Gas Sensors for Sensitive Room-Temperature Hydrogen and Carbon Monoxide Detection.* <https://doi.org/10.1021/acssensors.1c02015>
- Kuretake, T., Kawahara, S., Motoooka, M., & Uno, S. (2017). *An Electrochemical Gas Biosensor Based on Enzymes Immobilized on Chromatography Paper for Ethanol.* <https://doi.org/10.3390/s17020281>
- Lee, J., & Seo, M. (2022). *Palladium-based Electrical and Optical Hydrogen Gas Sensors.* 31(6), 397–402.
- Li, J. (2016). The electronic , structural and magnetic properties of oxygen vacancy : a first principles investigation. *Nature Publishing Group, March*, 1–14. <https://doi.org/10.1038/srep22422>
- Li, S., Guan, A., Wang, H., Yan, Y., Huang, H., Jing, C., Zhang, L., Zhang, L., & Zheng, G. (2022). *Hybrid palladium nanoparticles and nickel single atom catalysts for efficient electrocatalytic ethanol.* 6129–6133. <https://doi.org/10.1039/d1ta08518b>
- Li, T., Yin, W., Gao, S., Sun, Y., Xu, P., Wu, S., Kong, H., & Yang, G. (2022). The Combination of Two-Dimensional Nanomaterials with Metal Oxide Nanoparticles for Gas Sensors : A Review. *Nanomaterials*, 12(982). 10.3390/nano12060982
- Li, W., Guo, J., Cai, L., Qi, W., Sun, Y., Long Xu, J., Sun, M., Zhu, H., Xiang, L., Xie, D., & Ren, T. (2019). *UV Light irradiation enhanced gas sensor*

- selectivity of NO₂ and SO₂ using rGO functionalized with hollow SnO₂ nanofibers.* <https://doi.org/10.1016/j.snb.2019.03.133>
- Li, W., Liu, Z., Wu, W., & Li, S. (2023). Molecular understanding of the impacts of structural characteristics on ethanol adsorption performance for adsorption heat pumps. *Royal Society of Chemistry* (6). <https://doi.org/10.1039/d2me00222a>
- Li, X., Cui, H. L., Zhang, R., & Li, S. (2020). First-principles study of biaxial strain effect on NH₃ adsorbed Ti₂CO₂ monolayer. *Vacuum*, 179. <https://doi.org/10.1016/j.vacuum.2020.109574>
- Li, X., Xu, C., Du, X., Wang, Z., Huang, W., Sun, J., Wang, Y., & Li, Z. (2022). Assembled Reduced Graphene Oxide/Tugsten Diselenide/Pd Heterojunction with Matching Energy Bands for Quick Banana Ripeness Detection. *Foods*, 11(3):1879. Doi : <https://doi.org/10.3390%2Ffoods11131879>
- Liu, H., Zhu, D., Liu, W., Miao, T., Chen, J., Cheng, B., Qin, H., & Hu, J. (2023). Study on gas sensing characteristics of LaFeO₃ sensor under multi-wavelength light illumination. *Materials Letters*, 344. <https://doi.org/10.1016/j.matlet.2023.134350>
- Liu, L., Zhao, X., Ding, G., Han, C., & Liu, J. (2023). Fe₃N sites anchored reduced graphene oxide activate peroxymonosulfate via singlet oxygen dominated process: Performance and mechanisms. *Chemical Engineering Journal*, 470, 143820. Doi : <https://doi.org/10.1016/j.cej.2023.143820>
- Majhi, S. M., Mirzaei, A., Kim, H. W., & Kim, S. S. (2021). *Nanofiber Gas Sensors : An Overview.*
- Makkar, P., & Ghosh, N. N. (2021). *A review on the use of DFT for the prediction of the properties of nanomaterials.* 27897–27924. <https://doi.org/10.1039/d1ra04876g>
- Melia, R., & Elvaswer. (2022). Karakterisasi Arus-Tegangan Sensor Gas Hidrogen Dari Bahan Semikonduktor Heterokontak ZnO/SnO₂ (TiO₂). *Jurnal Fisika Unand*, 11(2), 263–270. <https://doi.org/10.25077/jfu.11.2.263->

270.2022

- Meng, F., Qin, L., Gao, H., Zhu, H., & Yuan, Z. (2024). Perovskite-structured LaFeO₃ modified In₂O₃ gas sensor with high selectivity and ultra-low detection limit for 2-butanone. *Journal of Alloys and Compounds*, 970(September 2023), 172464. <https://doi.org/10.1016/j.jallcom.2023.172464>
- Minezaki, T., Krüger, P., Annanouch, F. E., Casanova-ch, J., Alagh, A., Villar-garcia, I. J., Virginia, P., Llobet, E., & Bittencourt, C. (2023). *Hydrogen Sensing Mechanism of WS₂ Gas Sensors Analyzed with DFT and NAP-XPS*.
- Mirandha, A. (2016). *Efektivitas Limbah Media Tumbuh Jamur (Baglog) dengan Enkapsulasi Alginate Gel dalam Mengadsorpsi Ion Logam Kadmium*. 4–18.
- Mourik, T. Van, Bühl, M., & Gaigeot, M.-P. (2014). *Density functional theory across chemistry , physics and biology Subject Areas :*
- Moon, D. B., Bag, A., Lee, H.B., Meeseep, M., Lee, D.H., & Lee, N.E. (2021). A stretchable, room-temperature operable, chemiresistive gas sensor using nanohybrids of reduced graphene oxide and zinc oxide nanorods.
- Muktadir, M. G., Alam, A., Piya, A. A., & Shamim, S. U. D. (2022). *Exploring the adsorption ability with sensitivity and*. 29569–29584. <https://doi.org/10.1039/d2ra04011e>
- Nasri, A., Pétrissans, M., Fierro, V., & Celzard, A. (2021). *Gas sensing based on organic composite materials : Review of sensor types , Progresses and Challenges*. 1–57.
- Nikolic, M. V., Vasiljevic, Z. Z., Milovanovic, V., & Stamenkovic, Z. (2020). *Semiconductor Gas Sensors: Materials, Technology, Design, and Application*. 20(6694). <http://dx.doi.org/10.3390/s20226694>
- Numin, M. S., Hassan, A., Jumbri, K., Wng, K. K., Borhan, N., Daud, N. M. R.N.M., Nor A, M. A., Suhor, F., & Wahab, R, A. (2022). *A recent review on theoretical studies of Gemini surfactat corrosion inhibitors*

- Nundy, S., Ghos, A., Nath, R., Paul, A., Tahir, A., & Mallick, T. (2021). Reduced graphene oxide (rGO) aerogel: Efficient adsorbent for the elimination of antimony (III) and (V) from wastewater. *Journal of Hazardous Materils*, 420. <https://doi.org/10.1016/j.jhazmat.2021.126554>
- Ogoshi, E., Popolin, M., Mera, C., Gabriel, A., João, M. N., Oliveira, O. N., Fernando, J., & Gustavo, V. P. (2024). Learning from machine learning : the case of band - gap directness in semiconductors. *Discover Materials*, 4(6). <https://doi.org/10.1007/s43939-024-00073-x>
- Omoniyi, K. I., Aroh, A. O., Gimba, C. E., Abba, H., & Yilleng, M. T. (2018). *Determination of the Band Gap and Intensity of Palladium / Silver Doped TiO₂ Nano Particles Using Diffuse Reflectance and Photoluminescence Spectra*. 10(3), 60–66.
- Onyekwelu, K. (2019). *Ethanol*. January. <https://doi.org/10.5772/intechopen.79861>
- Peng, K., Fu, L., Yang, H., & Ouyang, J. (2016). Perovskite LaFeO₃ / montmorillonite nanocomposites : synthesis , interface characteristics and enhanced photocatalytic activity. *Nature Publishing Group, December 2015*, 1–10. <https://doi.org/10.1038/srep19723>
- Perez, M. P. M., Ortin, A. B. B., Porras, P. P., Jurado, R., & Plaza, E. G. (2020). *A New Approach to the Reduction of Alcohol Content in Red Wines : The Use of High-Power Ultrasounds*. 1–17.
- Pitriana, P., Wungu, T. D. K., Hidayat, R., & Herman, H. (2019). *Ab-initio calculation of APbI₃ (A = Li , Na , K , Rb and Cs) perovskite crystal and their lattice constants optimization using density functional theory Ab-initio calculation of APbI₃ (A = Li , Na , K , Rb and Cs) perovskite crystal and their lattic*. 3. <https://doi.org/10.1088/1742-6596/1170/1/012023>
- Purnamasari, I. (2021). Analisis Sifat Struktur dan Sifat Listrik pada Material Perovskite LaFe0.97Zr0.03O₃: XRD, Raman Scattering, SEM dan Impedansi Spectroskopi. *Kontruksi*, 19(1), 231–240.

- Qin, J., Cui, Z., Yang, X., Zhu, S., Li, Z., & Liang, Y. (2015). *Synthesis of three-dimensionally ordered macroporous LaFeO₃ with enhanced methanol gas sensing properties*. <https://doi.org/10.1016/j.snb.2014.12.046>
- Qin, W., Zhang, R., Yuan, Z., et al. (2022). Exposure Surface Active Sites of Perovskite-Type LaFeO₃ Gas Sensors by Selectively Dissolving La Cations for Enhancing Gas Sensing Properties to Acetone. *Advanced Materials Technologies*, 7(11). Doi: <https://doi.org/10.1002/admt.202200255>
- Rai, A., Thakur, A. K. (2017). *Effect of co-substitution on structural, optical, dielectric, and magnetic behavior of LaFeO₃*. <https://doi.org/10.1016/j.jallcom.2016.11.407>
- Raju, P., & Li, Q. (2022). Review — Semiconductor Materials and Devices for Gas Sensors Review — Semiconductor Materials and Devices for Gas Sensors. *Journal of The Electrochemical Society*. <https://doi.org/10.1149/1945-7111/ac6e0a>
- Ratner, M. H., Ewing, W. M., Rutchik, J. S., Ratner, M. H., Ewing, W. M., & Rutchik, J. S. (2020). Neurological effects of chronic occupational exposure to alcohol mists and vapors in a machinist. *Toxicology Communications*, 4(1), 43–48. <https://doi.org/10.1080/24734306.2020.1768341>
- Rattan, S., Kumar, S., & Goswamy, J. K. (2022). Gold nanoparticle decorated graphene for efficient sensing of NO₂ gas. *Sensors International*, 3(2), 100147. <https://doi.org/10.1016/j.sintl.2021.100147>
- Ren, K., Miao, J., Shen, W., Su, H., Pan, Y., Zhao, J., Pan, X., Li, Y., Fu, Y., Zhang, L., & Han, S. (2022). Progress in Natural Science: Materials International High temperature electrochemical discharge performance of LaFeO₃ coated with C / Ni as anode material for NiMH batteries. *Progress in Natural Science: Materials International*, 32(6), 684–692. <https://doi.org/10.1016/j.pnsc.2022.11.001>
- Renaldi, U., Baehaqi, M., & Wachyudin, D. (2021). *Rancang Bangun Alat Pendekripsi Kadar Alkohol pada Parfum Menggunakan Sensor MQ-3: Studi*

- Kasus pada Berbagai Merek Parfum.* 3(01).
- Rong, Q., Zhang, Y., Hu, J., Li, K., Wang, H., Chen, M., & Lv, T. (2018). Design of ultrasensitive Ag-LaFeO₃ methanol gas sensor based on quasi molecular imprinting technology. *Scientific Reports*, 8(14220), 1–12. <https://doi.org/10.1038/s41598-018-32113-x>
- Rostika, A., Noviyanti, A. R., Yusuf, A., Hartati, Y. W., Adiperdana, B., Padjadjaran, U., Padjadjaran, U., & Padjadjaran, U. (2022). *Studi Interaksi Gas Karbondioksida..... Atiek Rostika, dkk.* 57–65.
- Salehabadi, A., Enhessari, M., & Gupta, B. D. (2023) *Metal Chalcogenide Biosensors*.
- Saruhan, B., Fomekong, R. L., & Nahirniak, S. (2021). Review : Influences of Semiconductor Metal Oxide Properties on Gas Sensing Characteristics. *Fontier in Structure*, 2, 1–24. <https://doi.org/10.3389/fsens.2021.657931>
- Schmidt, N., & Stohr, M. (2018). *Molecular Self-Assembly on Graphene : The Role of the Substrate*. <https://doi.org/10.1016/B978-0-12-409547-2.14162-9>
- Shahdeo, D., Roberts, A., & Abbineni, N. (2020). Graphene based sensors. In *Analytical Applications of Graphene for Comprehensive Analytical Chemistry* (1st ed., Vol. 91). Elsevier B.V. <https://doi.org/10.1016/bs.coac.2020.08.007>
- Sharma, N., Kushwaha, H. S., Sharma, S. K., & Sachdev, K. (2020). Fabrication of LaFeO₃ and rGO-LaFeO₃ microspheres based gas sensors for detection of NO₂ and CO. *RSC Advance*, 10, 1297–1308. <https://doi.org/10.1039/c9ra09460a>
- Souri, M., Yamini, Y., & Amoli, H. S. (2023). The Synergistic effect of Ce dopant/cotton bio-template on the performance of the SnO₂ gas sensor for the detection of Ethanol. *Materials science and Engineering : B*, 294. <https://doi.org/10.1016/j.mseb.2023.116501>
- Suhendi, E., Haryadi, H., & Suprayoga, E. (2022). An Analysis of Electronic Properties of LaFeO₃ using Density Functional Theory with Generalized

- Gradient Approximation-Perdew-Burke-Ernzerhof Method for Ethanol Gas Sensors. *Materials Research*. 2022; 25:E20210554. <https://doi.org/10.1590/1980-5373-MR-2021-0554%0AAn>
- Surendra Babu, N. (2021). Applications of Current Density Functional Theory (DFT) Methods in Polymer Solar Cells. *InTechOpen*. doi: 10.5772/intechopen.100136
- Tangirala, V. K. K., Pozos, H. G., Lugo, V. R., & Olvera, M. D. L. luz. (2017). *A Study of the CO Sensing Responses of Cu-, Pt- and*. <https://doi.org/10.3390/s17051011>
- Theingi, M., Tun, K. T., & Aung, N. N. (2019). *Preparation , Characterization and Optical Property of LaFeO₃ Nanoparticles via Sol-Gel Combustion Method*. 1(3), 151–157.
- Tshwane, D. M., & Modiba, R. (2023). Surface Science Oxidation and corrosion investigation on Ti 2 AlV (110) surface using first principle approach ☆. *Surface Science*, 737(July), 122367. <https://doi.org/10.1016/j.susc.2023.122367>
- Waghmare, Y. A., Narwade, V. N., Umar, A., Ibrahim, A. A., & Shirsat, M. D. (2024). Enhanced CO sensing with highly sensitive and selective rGO-Ru OEP chemiresistive sensor. *Chemical Physics Impact*, 8(October 2023), 100419. <https://doi.org/10.1016/j.chphi.2023.100419>
- Wang, C., Rong, Q., Zhang, Y., Hu, J., Zi, B., Zhu, Z., Zhang, J., & Liu, Q. (2019). *Molecular imprinting Ag-LaFeO₃ Spheres for highly sensitive acetone gas detection*
- Wang, J., Shen, T., Feng, Y., & Liu, H. (2020). *Physica B : Physics of Condensed Matter* A GGA þ U study of electronic structure and the optical properties of different concentrations Tb doped ZnO. *Physica B: Physics of Condensed Matter*, 576(August 2019), 411720. <https://doi.org/10.1016/j.physb.2019.411720>

- Wang, P., Zhan, S., Xia, Y., Ma, S., Zhou, Q., & Li, Y. (2017). The fundamental role and mechanism of reduced graphene oxide in rGO/Pt-TiO₂ nanocomposite for high-performance photocatalytic water splitting. *Applied Catalysis B: Environmental*, 207. <https://doi.org/10.1016/j.apcatb.2017.02.031>
- Wang, T., Sun, P., Liu, F., & Lu, G. (2022). Reveaking the correlation between gas selectivity and semiconductor energy band structure derived from off-stoichiometric spinel CdGa₂O₄.
- Wang, X., Qin, H., Pei, J., Chen, Y., Li, L., Xie, J., & Hu, J. (2016). Sensing performance to low concentration acetone for palladium doped LaFeO₃ sensors. [https://doi.org/10.1016/S1002-0721\(16\)60082-0](https://doi.org/10.1016/S1002-0721(16)60082-0)
- Widyanti, E. M., & Moehadi, I. (2016). *Proses Pembuatan Etanol Dari Gula Menggunakan Saccharomyces Cerevisiae Amobil*. 12(2), 31–38. <http://ejournal.undip.ac.id/index.php/metana>
- Woods, N. (2018). *The University of Cambridge*.
- Wusnah, Bahri, S., & Hartono, D. (2016). PROSES PEMBUATAN BIOETANOL DARI KULIT PISANG KEPOK (*Musa acuminata* B.C) SECARA FERMENTASI. *Jurnal Teknologi Kimia Unimal*, 1, 57–65.
- Yokoyama, M., Nakada, K., & Ishii, A. (2014). Density functional theory calculations for Pd adsorption on SO₄ adsorbed on h-BN. *Computational Materials Science*, 82, 231–236. <https://doi.org/10.1016/j.commatsci.2013.08.058>
- Yu, J., Wang, C., Yuan, Q., Yu, X., & Wang, D. (2022). Ag-Modified Porous Perovskite-Type LaFeO₃ for Efficient Ethanol Detection. *Nanomaterials*, 12(1768), 1–13. <https://doi.org/https://doi.org/10.3390/nano12101768>
- Yuan, C., Ma, J., Zou, Y., Li, G., Xu, H., & Sysoev, V. V. (2022). Modeling Interfacial Interaction between Gas Molecules and Semiconductor Metal Oxides: A New View Angle on Gas Sensing. 2203594, 1–35. <https://doi.org/10.1002/advs.202203594>

- Yuan, G., Zhong, Y., Chen, Y., & Sun, X. (2022). *RSC Advances Highly sensitive and fast-response ethanol sensing of porous Co₃O₄ hollow polyhedra via palladium reined spillover effect*. 6725–6731. <https://doi.org/10.1039/d1ra09352e>
- Zabelin, D., Zabelina, A., Guselnikova, O., Miliutina, E., Kolska, Z., Stulik, J., Polansky, R., Elashnikov, R., Kalachyova, Y., Svorcik, V., & Lyutakov, O. *Selective methane chemiresistive detection using MWCNTs array decorated by metal organic framework layer.* <https://doi.org/10.1016/j.surfin.2023.103105>
- Zhang, H., Wang, J., Zhou, Y., Tu, C., Fu, C., dkk. (2018). Occurrences of Organophosphorus Esters and Phthalates in the Microplastic from the Coastal Beaches in north China.
- Zhang, H., Xiao, J., Chen, J., Wang, Y., Zhang, L., Yue, S., Li, S., & Huang, T. (2022). *Pd-Modified LaFeO₃ as a High-Efficiency Gas-Sensing Material for H₂S Gas Detection.* 12(14), 2460. <https://doi.org/10.3390/nano12142460>
- Zhang, R., Zhao, C., Huo, Y., Han, Y., Hong, J., Liu, Y., Zhang, A., Guo, R., & Ai, Y. (2022). *Theoretical calculation of toxix/radioactive metal ion capture by novel nanomaterial*
- Zhang, X., Sun, J., Tang, K., Wang, H., Chen, T., & Jiang, K. (2022). *Ultralow detection limit and ultrafast response / recovery of the H₂ gas sensor based on Pd-doped rGO / ZnO-SnO₂ from hydrothermal synthesis.* <https://doi.org/10.1038/s41378-022-00398-8>
- Zhou, Y., Lü, Z., Li, J., Xu, S., Xu, D., & Wei, B. (2021). The electronic properties and structural stability of LaFeO₃ oxide by niobium doping: A density functional theory study. *International Journal of Hydrogen Energy*, 46(13), 9193–9198. <https://doi.org/10.1016/j.ijhydene.2020.12.202>